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Microwave-Assisted Synthesis of Spiro(cycloalkane thiazolo-s-tetrazine)

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Microwave irradiation of cyclic ketones with thiocarbohydrazide furnish the corresponding s-tetrazine derivatives 1a–d, which upon irradiation with chloroacetic acid and sodium acetate in acetic acid, give the corresponding thiazolidinone derivatives 2a–d. These derivatives 2a–d upon condensation with benzaldehyde afford the benzylidine derivatives 3a–d. The s-tetrazine derivatives 1a–d and 1,2-dibromoethane upon irradiation gives the thiazolo-s-tetrazine derivatives 4a–d. The structures of 1–4 were established by IR, 1H NMR, and 13C NMR data.

Keywords Irradiation; microwave assisted; spiro; tetrazine

INTRODUCTION

Sometimes, in a chemical reaction, starting materials as well as products decompose or unwanted products are formed because of a long reaction time and high temperature, which invariably reduce the percentage of yield. Thus the production of a large amount of waste from a reaction is undesirable both from economic and an environmental point of view. Green technologies form an alternative method to allow a more clean and efficient chemical syntheses. Microwave energy is fast becoming the method of choice for both industrial and academic chemists for driving reactions to completion, as it offers the safest, most effective way to increase reaction rates and improve product yields, while promoting green chemistry ideas. Reactions that previously took hours, or

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even days, to complete can now be performed in minutes. 1-8 In the last decade, some of the first examples yielded interesting results, which led to much research in this novel technique. 9-19 Dandia et al. 20 reported a one-pot synthesis of novel spiro[pyrazolo[4,3-c][1, 5]benzothiazepines] with a 71% yield within 6-8 min using microwave irradiation. They also reported the synthesis of novel spiro[indole-pyrido[2,3-d]pyrimidines] by the reaction of in situ generated spiro[indole-dihydropyridine] and urea/CS2 using basic alumina as a solid support or a few drops of DMF as a homogenizer under microwave irradiation for 3-4 min in 85-89% vield. 21 Spiro-(indoline-isoxazolidines) were prepared in modest yields, by the cycloaddition reaction between ethyl(3-indolylidene)-acetate and various substituted α, N-diphenylnitrones under solvent-free conditions using a microwave method.²² The reaction conducted under conventional heating in an oil bath did not proceed even after 20 h, especially when it was carried out without solvent. The 4'-(2-butyl-4-oxo-1,3diaza-spiro[4.4]non-1-en-3-ylmethyl)-biphenyl-2-carbonitrile was synthesized by irradiating methyl 1-amino-1-cyclopentanecarboxylate and valeric acid in anhydrous pyridine, with triphenyl phosphate (neat).²³

In our ongoing research program on the development of spiroheterocycles, we became interested in the efficient and microwave-assisted syntheses of the spirotetrazines into which thiazole and thiazolidinone moieties were incorporated. Pujari and colleagues^{24–27} reported the syntheses of spirotetrazines derivatives (1–4) by conventional methods that required long reaction times and only moderate yields. Therefore, we have investigated the syntheses of these compounds (1–4) in microwave conditions.

RESULTS AND DISCUSSION

Cycloalkanones of different sizes undergo condensation with thiocabohydrazide under microwave condition to yield corresponding s-tetrazine derivatives 1a-d. Thiazolidinone derivatives 2a-d are obtained upon irradiation of a mixture of s-tetrazine derivatives 1a-d and chloroacetic acid in acetic acid in the presence of sodium acetate. These thiazolidinone derivatives 2a-d containing active methylene groups are subjected to condensation with benzaldehyde to form respective benzylidine compounds 3a-d. Thiazolo-s-tetrazine derivatives 4a-d are also obtained by the reaction of the s-tetrazine derivatives 1a-d with dibromoethane. These reactions are depicted in Scheme 1.

The IR spectral data of the compounds synthesized by microwave irradiation were identical to those reported earlier. ^{24–27} The ¹H NMR

SCHEME 1

spectral data of compounds **2–4** revealed some interesting results, which were not observed by Pujari and colleagues. $^{24-27}$ It was reported that the methylene protons of cycloalkane ring adjacent to the spiro carbon in compounds **2** appeared as one multiplet at δ 2.46–2.62 in their ¹H NMR spectra. In the case of these compounds synthesized by microwave irradiation, two clear multiplets were observed in the ¹H

NMR spectra at δ 2.54–2.57 and at δ 2.68–2.71. This is because these compounds have a thiazolidinone moiety on one side of the spirote-trazine structure, which makes the methylene proton of cycloalkane ring adjacent to spiro carbon slightly different. The downfield shift experienced by one of the methylene protons is due to the carbonyl group of thiazolidinone ring in **2**, which is evident from the fact that in case of compound **4**, where there is no carbonyl group, only one multiplet has been observed at δ 2.42–2.52.

The ¹H NMR data of compounds **3c** and **4c** and the ¹³C NMR data of compounds **1c–4c** are also given in the appropriate places in the Experimental section, which were not reported earlier. ²⁵ The ¹³C NMR spectral data provides strong evidence in favor of the formation of the products as mentioned and also confirm the proposed structures.

We have synthesized 16 novel spiro compounds. The IR and ¹H NMR data of all the compounds synthesized by microwave irradiation were found to be almost identical with those reported earlier. ^{24–27} The analytical and spectral data of compounds **3b**, **3d**, **4a**, **4c**, and **4d** synthesized by the microwave method are given in the Experimental section, as these were not reported earlier. The reaction times, yields of the products (1–4) under microwave condition and from the classical heating method, and also the melting points of these compounds are given in the Table I.

EXPERIMENTAL

The melting points were determined on a Zenith apparatus and are uncorrected. IR spectra were recorded on a Perkins Elmer FT-IR spectrophotometer using KBr disc. ¹H and ¹³C NMR spectra were recorded on a 400 MHz FT NMR spectrometer and were recorded in CDCl₃. Purity of the products was checked by TLC, using silica gel G (BDH) and toluene:ethyl acetate (1:1) as eluent. The reactions were done in a domestic microwave oven (LG make) operated at 700 Watt generating 2450 MHz frequency. Thiocarbohydrazide was prepared according to the reported procedure.²⁸

1,2,4,5-Tetraazaspiro[5.6]dodecane-3-thione (1c)

A mixture of thiocarbohydrazide (1.06 g, 0.01 mol) and cycloheptanone (1.18 mL, 0.01 mol) in water:ethanol (1:1, 2 mL) was taken in a beaker (25 mL). Then the mixture was irradiated in a microwave oven for 50 sec. The cream-colored precipitate obtained was washed well with hot water (25 mL) and finally crystallized from ethanol to obtain needleshaped crystals. Mp 155°C, Yield 95%, (Found S, 15.89, $C_8H_{16}N_4S$)

			Microwave method			Conventional method			
Comp. No.	n	Molecular formula	Mp (°C)	Yield (%)	Time (Sec)	Mp (°C)	Yield (%)	Time (h)	S% Cal. (Found)
1a	0	$C_6H_{12}N_4S$	167	92	50	179 ^a	89 ^a	24 ^a	21.62 (21.55)
1b	1	$C_7H_{14}N_4S$	168	93	40	169^{b}	$81^{\rm b}$	$24^{ m b}$	17.20 (17.16)
1c	2	$C_8H_{16}N_4S$	155	95	50	$150^{\rm c}$	85^{c}	24^{c}	16.00 (15.89)
1d	3	$C_9H_{18}N_4S$	175	90	45	$180^{ m d}$	$84^{ m d}$	$24^{ m d}$	14.95 (14.90)
2a	0	$C_8H_{12}N_4SO$	140	70	40	$144^{\rm e}$	$59^{\rm e}$	6^{e}	15.09 (14.96)
2b	1	$C_9H_{14}N_4SO$	129	75	40	$130^{ m b}$	$53^{ m b}$	$5^{ m b}$	14.16 (13.99)
2c	2	$C_{10}H_{16}N_4SO$	133	77	40	135^{c}	66^{c}	5^{c}	13.33 (13.18)
2d	3	$C_{11}H_{18}N_4SO$	125	75	40	120^{d}	63^{d}	$5^{ m d}$	12.60 (12.54)
3a	0	$C_{15}H_{16}N_4SO$	180	62	40	$177^{\rm e}$	$40^{\rm e}$	3^{e}	10.67 (10.61)
3c	2	$C_{17}H_{20}N_4SO$	235	70	40	240^{c}	$50^{\rm c}$	3^{c}	9.76 (9.68)
4b	1	$C_9H_{16}N_4S$	176	60	35	$179^{\rm b}$	$44^{\rm b}$	6^{b}	15.09 (14.95)

TABLE I Physical and Analytical Data

requires S, 16.0%). IR (KBr) in cm⁻¹: 1231 (C=S), 3155–3205 (N-H). ¹H NMR (CDCl₃): δ 1.64 (m, 8H, 4 × C \boldsymbol{H}_2), 2.35 (m, 4H, 2 × C \boldsymbol{H}_2), 4.35 (bs, 2H, 2 × N \boldsymbol{H}), 8.44, 8.50 (bs, 2H, 2 × N \boldsymbol{H} CS). ¹³C NMR (CDCl₃): δ 24.23, 27.50, 30.10, 36.88, 179.72 (C=S).

6'(7'H)-Oxospiro[cycloheptane-1,3'(4'H)-[2H]thiazolo[3,2-b]-s-tetrazine] (2c)

To a mixture of **1c** (2.00 g, 0.01 mol), chloroacetic acid (0.95 g, 0.01 mol), and fused sodium acetate (0.82 g, 0.01 mol), a few drops of glacial acetic acid were added and the mixture was irradiated in a microwave oven for 40 sec. Then the reaction mixture was poured into ice cold water, and the precipitate obtained was filtered, washed well with water (50 mL), and crystallized from ethanol. Mp 133°C, Yield 77%, (Found S, 13.18, $C_{10}H_{16}N_4SO$ requires S, 13.33%). IR (KBr) in cm⁻¹: 1613(C=N), 1738(C=O), 3209–3318 (N-H). ¹H NMR (CDCl₃): δ 1.64 (m, 8H, 4 × C H_2), 2.56 (m, 2H, C H_2), 2.71 (m, 2H, C H_2), 3.74 (s, 2H, S-C H_2), 4.59 (bs, 2H, 2 × NH). ¹³C NMR (CDCl₃): δ 24.24, 27.50, 30.12, 32.49 (Thiazolidinone CH_2), 36.88, 149.5 (N = C=S), 171.3 (C=O).

aRef.29

 $^{^{\}rm b}{
m Ref.}^{\rm 24}$

cRef.²⁶

dRef.27

 $^{^{}m e}{
m Ref.}^{25}$

7'-Benzylidine-6'(7'H)-oxospiro[cycloheptane-1,3'(4'H)-[2H]thiazolo[3,2-b]-s-tetrazine] (3c)

To a mixture of **2c** (0.24 g, 0.001 mol), benzaldehyde (0.1 mL, 0.001 mol), and sodium acetate (0.16 g, 0.002 mol), a few drops of glacial acetic acid were added and the mixture was irradiated in a microwave oven for 40 sec. The yellow-colored precipitate obtained was washed well with water (50 mL) and then crystallized from glacial acetic acid. Mp 235°C, Yield 70%, (Found S, 9.68, $C_{17}H_{20}N_4SO$ requires S, 9.76%). IR (KBr) in cm⁻¹: 1615 (C=N), 1705 (C=O), 3201–3315 (N-H). ¹H NMR (CDCl₃): δ 1.64 (m, 8H, 4 × C H_2), 2.54 (m, 2H, C H_2), 2.73 (m, 2H, C H_2), 4.57 (bs, 2H, 2 × NH), 7.71 (m, 6H, Ar-H & = CH). ¹³C NMR (CDCl₃): δ 24.22, 27.49, 30.12, 36.88, 122.2 (Thiazolidinone C = CH), 127.1, 128.6, 129.4, 132.5 (=CH), 136.8, 149.4 (N = C-S), 166.22 (C=O).

7'-Benzylidene-6'(7'H)-oxospiro[cyclohexane-1,3'(4')H-[2H]thiazole[3,2-b]-s-tetrazine] (3b)

It was prepared from **2b** by the same method as above in 40 sec. Mp 215°C, Yield 65%, (Found S, 10.19, $C_{16}H_{18}N_4SO$ requires S, 9.99%). IR (KBr) in cm⁻¹: 1565, 1630, 1705, 3240, 3300. ¹H NMR (CDCl₃): 1.55 (m, 6H, $3 \times CH_2$), 2.52 (m, 4H, $2 \times CH_2$), 4.59 (bs, 2H, $2 \times NH$), 7.74 (m, 6H, Ar-H & = CH). ¹³C NMR (CDCl₃): 24.22, 27.49, 30.12, 36.88, 122.2, 127.1, 128.6, 129.4, 132.5, 136.8, 149.4, 166.2.

7'-Benzylidene-6'(7'H)-oxospiro[cyclooctane-1,3'(4')H-[2H]thiazole[3,2-b]-s-tetrazine] (3d)

It was prepared from **2d** by the same method as above in 60 sec. Mp 225°C, Yield 64%, (Found S, 9.36, $C_{18}H_{22}N_4SO$ requires S, 9.24%). IR (KBr) in cm⁻¹: 1570, 1635, 1708, 3245, 3310. ¹H NMR (CDCl₃): 1.58 (m, 10H, $5 \times CH_2$), 2.46 (m, 4H, $2 \times CH_2$), 4.60 (bs, 2H, $2 \times NH$), 7.69 (m, 6H, Ar-H & = CH). ¹³C NMR (CDCl₃): 23.60, 24.34, 27.55, 30.28, 37.10, 122.8, 127.4, 128.8, 130.2, 133.0, 137.4, 150.6, 167.1.

6',7'-Dihydrospiro[cycloheptane-1,3'(4'H)-[2H]thiazolo[3,2-b]-s-tetrazine] (4c)

To a mixture of 1c (2.00 g, 0.01 mol) and 1,2-dibromoethane (1.88 g, 0.01 mol), a few drops of glacial acetic acid were added and the mixture was irradiated in a microwave oven for 35 sec. Then the reaction mixture was cooled, ice cold water was added, and the precipitate obtained was filtered, dried, and crystallized from ethanol. Mp 170°C, Yield 74%,

(Found S, 14.09, $C_{10}H_{18}N_4S$ requires S, 14.16%). IR (KBr) in cm⁻¹: 1610 (C=N), 3160–3250 (N-H). ¹H NMR (CDCl₃): δ 1.67 (m, 8H, 4 × C \boldsymbol{H}_2), 2.50 (m, 4H, 2 × C \boldsymbol{H}_2), 3.16 (t, 2H, S-C \boldsymbol{H}_2), 3.85 (t, 2H, -N-C \boldsymbol{H}_2), 4.45 (bs, 2H, 2 × N \boldsymbol{H}). ¹³C NMR (CDCl₃): δ 24.24, 27.48, 30.12, 36.12 (S- $\boldsymbol{C}H_2$), 36.88, 55.64 (N- $\boldsymbol{C}H_2$), 149.4 (N= \boldsymbol{C} -S).

6',7'-Dihydrospiro[cyclopentane-1,3'(4'H)-[2H]thiazolo[3,2-b]-s-tetrazine] (4a)

It was prepared from **1a** by the same method as above in 35 secs. Mp 175°C, Yield 62%, (Found S, 16.02, $C_8H_{14}N_4S$ requires S, 16.16%). IR (KBr) in cm⁻¹: 1530, 1600, 3080, 3160. ¹H NMR (CDCl₃): δ 1.72 (m, 4H, 2 × CH₂), 2.55 (m, 4H, 2 × CH₂), 3.20 (t, 2H, S-CH₂), 3.88 (t, 2H, -N-CH₂), 4.52 (bs, 2H, 2 × NH). ¹³C NMR (CDCl₃): δ 27.60, 30.38, 36.44, 37.15, 55.74, 150.10.

6',7'-Dihydrospiro[cyclooctane-1,3'(4'H)-[2H]thiazolo[3,2-b]-s-tetrazine] (4d)

It was prepared from **1d** by the same method as above in 35 sec. Mp 171°C, Yield 69%, (Found S, 13.33, $C_{11}H_{20}N_4S$ requires S, 13.14%). IR (KBr) in cm⁻¹: 1560, 1610, 3160–3250. ¹H NMR (CDCl₃): δ 1.58 (m, 10H, 5 × CH₂), 2.52 (m, 4H, 2 × CH₂), 3.18 (t, 2H, S-CH₂), 3.86 (t, 2H, -N-CH₂), 4.64 (bs, 2H, 2 × NH). ¹³C NMR (CDCl₃): δ 23.58, 24.44, 27.56, 30.32, 36.12, 36.90, 55.70, 149.8.

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